MASS SPECTROMETRIC FRAGMENTATION OF TRITERPENOID DERIVATIVES WITH OXABICYCLOOCTANE AND OXABICYCLOHEPTANE RING E ARRANGEMENT*

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The character of the mass spectra of triterpenoids derived from 20β , 28-epoxy-18 α , 19 β H-ursane (I) is determined by the structure of the ring E. Fragmentation always takes place in ring E. Two types of fragmentation were found during which ring D is either preserved or cleaved. The type of fragmentation depends on the substitution at the positions $C_{(21)}$ and $C_{(22)}$.

In the preceding paper¹ we discussed the structure elucidation and the reactivity of derivatives of 20β , 28-epoxy-18 α , 19 β H-ursane (I). In connection with structure determination of these compounds the mass spectra of compounds II - XVIII were studied and the results of these measurements are summarized in this paper (Table I). All compounds measured contain in their ring E an oxabicyclo[2,2,2]octane or oxabicyclo[2,2,1]heptane system (anhydride II of oxabicyclo[3,2,2]nonane), part of which is always the same tetrahydropyran ring *cis* annelated with ring D. The individual compounds differ only by the bridge over this tetrahydropyran ring between the positions 17 and 20; this bridge determines the character of the fragmentation.

EXPERIMENTAL

The measurement was carried out on a Varian MAT 311 spectrometer. The energy of the ionizing electrons was 70 eV and the ionizing electron current was 1 mA; the temperature of the ion source was 200° C and the temperature of the direct inlet system was $150-200^{\circ}$ C. The high resolution measurements were carried out with an error not exceeding 5 p.p.m.

RESULTS AND DISCUSSION

The compounds measured can be divided into two groups depending on the type of bridging of the tetrahydropyran ring at the positions $C_{(17)}$ and $C_{(20)}$. The type A (compounds II - X): The group forming the bridge is eliminated in the form of a neutral molecule (or neutral molecules), while ring D remains uncleaved (Scheme 1).

^{*} Part L in the series Triterpenes; Part IL: This Journal 41, 1200 (1976).

Fragmentation of Triterpenoid Derivatives

TABLE I

Ions of the Substances Measured

Small letters indicate the type of ion, see Schemes 1 and 2.

- *II* (Anhydride of 3β-acetoxy-21,22-seco-20β,28-epoxy-18α,19β*H*-ursane-21,22-dioic acid): m/e 528 (C₃₂H₄₈O₆, 0·1%, M), 500 (C₃₁H₄₈O₅, 1·6%, b), 456 (C₃₀H₄₈O₃, 15·0%, f), 440 (C₂₉H₄₄O₃, 8·7%, b), 425 (C₂₈H₄₁O₃, 7·9%, 440-CH₃), 396 (C₂₈H₄₄O, 3·5%, f), 384 (C₂₆H₄₀O₂, 30·7%, g), 324 (C₂₄H₃₆, 7·9%, g), 189 (C₁₄H₂₁, 52·8%)^a, 43 (100%)^b
- III (3β-Hydroxy-20β,28-epoxy-21-oxa-18α,19βH-ursan-22-one): m/e 458 (C₂₉H₄₆O₄, 5·8%, M), 440 (C₂₉H₄₄O₃, 5·2%, b), 425 (C₂₈H₄₁O₃, 7·0%, 440-CH₃), 414 (C₂₈H₄₆O₂, 12·8%, f), 396 (C₂₈H₄₄O, 7·0%, f), 342 (C₂₄H₃₈O, 32·6%, g), 324 (C₂₄H₃₆, 7·6%, g), 189 (C₁₄H₂₁, 59·3%), 43 (100%)
- *IV* (3β-Acetoxy-20β,28-epoxy-21-oxa-18α,19β*H*-ursan-22-one): m/e 500 (C₃₁H₄₈O₅, 1·0%, M), 456 (C₃₀H₄₈O₃, 15·2%, *f*), 440 (C₂₉H₄₄O₃, 9·5%, *b*), 425 (C₂₈H₄₁O₃, 8·6%, 440-CH₃), 396 (C₂₈H₄₄O, 2·9%, *f*), 384 (C₂₆H₄₀O₂, 28·6%, *g*), 324 (C₂₄H₃₆, 6·2%, *g*), 189 (C₁₄H₂₁, 47·1%), 43 (100%)
- $V \quad (20\beta,28 \text{-epoxy-21-oxa-18}\alpha,19\beta H \text{-urs-2-en-22-one}): \ m/e \ 440 \ (C_{29}H_{44}O_3, \ 10.8\%, \ M), \\ 425 \ (C_{28}H_{41}O_3, \ 12.8\%, \ 440 \text{-}CH_3), \ 396 \ (C_{28}H_{44}O, \ 22.1\%, f), \ 324 \ (C_{24}H_{36}, \ 41.2\%, g), \\ 189 \ (C_{14}H_{21}, \ 60.8\%), \ 43 \ (100\%)$
- *VI* (3β-Hydroxy-20β,28-epoxy-*E*(21)-nor-18α,19β*H*-ursan-22-one): m/e 442 (C₂₉H₄₆O₃, 4·4%, M), 414 (C₂₈H₄₆O₂, 27·9%, *f*), 396 (C₂₈H₄₄O, 15·2%, *f*), 342 (C₂₄H₃₈O, 54·4%, *g*), 324 (C₂₄H₃₆, 7·6%, *g*), 189 (C₁₄H₂₁, 38·0%), 43 (100%)
- *VII* (3β-Acetoxy-20β,28-epoxy-*E*(21)-nor-18α,19β*H*-ůrsan-22-one): m/e 484 (C₃₁H₄₈O₄, 0·1%, M), 456 (C₃₀H₄₈O₃, 15·3%, *f*), 396 (C₂₈H₄₄O, 5·9%, *f*), 384 (C₂₆H₄₀O₂, 24·7%, *g*), 324 (C₂₄H₃₆, 11·8%, *g*), 189 (C₁₄H₂₁, 56·5%), 43 (100%)
- VIII (3β,22ξ-Dihydroxy-20β,28-epoxy-E(21)-nor-18α,19βH-ursane-22ξ-carboxyloic acid): m/e
 488 (C₃₀H₄₈O₅, 4·4%, M), 414 (C₂₈H₄₆O₂, 6·0%, f), 396 (C₂₈H₄₄O, 5·5%, f), 342 (C₂₄H₃₈O, 11·7%, g), 324 (C₂₄H₃₆, 4·4%, g), 189 (C₁₄H₂₁, 20·1%), 43 (100%)
- *IX* (3β,22ξ-Diacetoxy-20β,28-epoxy-*E*(21)-nor-18α,19β*H*-ursane-22ξ-carboxyloic acid): m/e572 (C₃₄H₅₂O₇, 0·3%, M), 456 (C₃₀H₄₈O₃, 8·2%, *f*), 396 (C₂₈H₄₄O, 2·3%, *f*), 384 (C₂₆H₄₀O₂, 8·8%, *g*), 324 (C₂₄H₃₆, 3·8%, *g*), 189 (C₁₄H₂₁, 22·0%), 43 (100%)
- X (Methyl 3β-acetoxy-22ξ-hydroxy-20β,28-epoxy-E(21)-nor-18α,19β*H*-ursane-22ξ-carboxylate): m/e 544 (C₃₃H₅₂O₆, 43.6%, M), 484 (C₃₁H₄₈O₄, 23.1%, c), 456 (C₃₀H₄₈O₃, 68.0%, f), 396 (C₂₈H₄₄O, 34.6%, f), 384 (C₂₆H₄₀O₂, 83.3%, g), 324 (C₂₄H₃₆, 42.3%, g), 189 (C₁₄H₂₁, 91.7%), 43 (100%)
- XI (20 β ,28-Epoxy-*E*(21)-nor-18 α ,19 β *H*-ursane-3 β ,22 α -diol): *m/e* 444 (C₂₉H₄₈O₃, 100%, M), 426 (C₂₉H₄₆O₂, 8·1%, *j*), 301 (C₂₂H₃₇, 8·8%, *n*), 189 (C₁₄H₂₁, 18·9%), 125 (C₇H₉O₂, 82·4%, *o*), 43 (45·9%)
- XII (3β-Acetoxy-20β,28-epoxy-E(21)-nor-18 α ,19βH-ursan-22 α -ol): m/e 486 (C₃₁H₅₀O₄, 56·7%, M), 426 (C₂₉H₄₆O₂, 10·3%, j), 301 (C₂₂H₃₇, 12·4%, n), 189 (C₁₄H₂₁, 28·9%), 125 (C₇H₉O₂, 100%, o), 43 (64·9%)

TABLE I

(Continued)

- XIII (3 β ,22 α -Diacetoxy-20 β ,28-epoxy-E(21)-nor-18 α ,19 β H-ursane): m/e 528 (C₃₃H₅₂O₅, 35·5%, M), 486 (C₃₁H₅₀O₄, 56·5%, j), 468 (C₃₁H₄₈O₃, 8·1%, i), 426 (C₂₉H₄₆O₂, 10·5%, j), 301 (C₂₂H₃₇, 11·3%, n), 189 (C₁₄H₂₁, 49·2%), 125 (C₇H₉O₂, 87·1%, o), 43 (100%)
- XIV (3β-Acetoxy-20β,28-epoxy-E(21)-nor-18α,19βH-ursan-22β-ol): m/e 486 (C₃₁H₅₀O₄, 94·4%, M), 426 (C₂₉H₄₆O₂, 3·3%, j), 301 (C₂₂H₃₇, 11·1%, n), 189 (C₁₄H₂₁, 26·7%), 125 (C₇H₉O₂, 100%, o), 43 (96·7%)
- XV (3 β ,22 β -Diacetoxy-20 β ,28-epoxy-E(21)-nor-18 α ,19 β H-ursane): m/e 528 (C₃₃H₅₂O₅, 51·3%, M), 486 (C₃₁H₅₀O₄, 60·0%, j), 468 (C₃₁H₄₈O₃, 3·8%, i), 426 (C₂₉H₄₆O₂, 3·8%, j), 301 (C₂₂H₃₇, 7·5%, n), 189 (C₁₄H₂₁, 25·0%), 125 (C₇H₉O₂, 48·8%, o), 43 (100%)
- XVI (22 ξ -Hydroxymethyl-20 β ,28-epoxy-E(21)-nor-18 α ,19 β H-ursane-3 β ,22 ξ -diol): m/e 474 (C₃₀H₅₀O₄, 1.5%, M), 456 (C₃₀H₄₈O₃, 16.2%, k), 438 (C₃₀H₄₆O₂, 11.1%, l), 410 (C₂₉H₄₆O, 12.1%, m), 395 (C₂₈H₄₃O, 12.1%, 410-CH₃), 301 (C₂₂H₃₇, 9.1%, n), 189 (C₁₄H₂₁, 90.9%), 109 (C₇H₉O, 100%, p), 43 (66.7%)
- XVII (3β-Acetoxy-22ξ-acetoxymethyl-20β,28-epoxy-E(21)-nor-18α,19βH-ursan-22ξ-ol): m/e558(C₃₄H₅₄O₆, 0·2%, M), 498 (C₃₂H₅₀O₄, 34·3%, k), 438 (C₃₀H₄₆O₂, 37·3%, l), 410 (C₂₉H₄₆O, 2·9%, m), 395 (C₂₈H₄₃O, 26·5%, 410-CH₃), 301 (C₂₂H₃₇, 5·9%, n), 189 (C₁₄H₂₁, 71·6%), 109 (C₇H₉O, 56·9%, p), 43 (100%)
- XVIII (3β-Acetoxy-20β,28-epoxy-18α,19βH-ursan-21-one): m/e 498 ($C_{32}H_{50}O_4$, 0.7%, M), 470 ($C_{31}H_{50}O_3$, 1.8%, m), 438 ($C_{30}H_{46}O_2$, 0.7%, *l*), 410 ($C_{29}H_{46}O$, 0.8%, m), 395 ($C_{28}H_{43}O$, 1.7%, 410-CH₃), 361 ($C_{24}H_{41}O_2$, 1.1%, n), 301 ($C_{22}H_{37}$, 9.0%, n), 189 ($C_{14}H_{21}$, 7.9%), 109 (C_7H_9O , 100%, p), 43 (25.8%)

^{*a*} Ion characteristic of pentacyclic triterpenoids². ^{*b*} Mixture of ions C_3H_7 , C_2H_3O from ring E of compounds of type A and C_2H_3O from ring A of 3β-acetoxy derivatives.

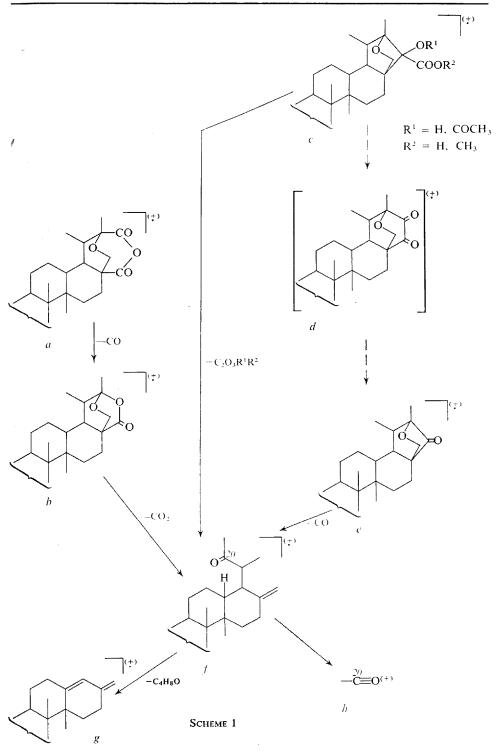
Type B (compounds XI-XVIII): The group forming the bridge is not eliminated and ring D is cleaved (Scheme 2). The following groups are eliminated: --CO--O--CO--, --O--CO--, --C(OH)(COOH)--, --C(OCOCH_3). .(COOH)-- and --C(OH)(COOCH_3)--. The following groups are not eliminated: --CH(OH)-, --CH(OCOCH_3), --C(OH)(CH_2OH)--, --C(OH)(CH_2OCOCH_3)--, and --COCH_2--.

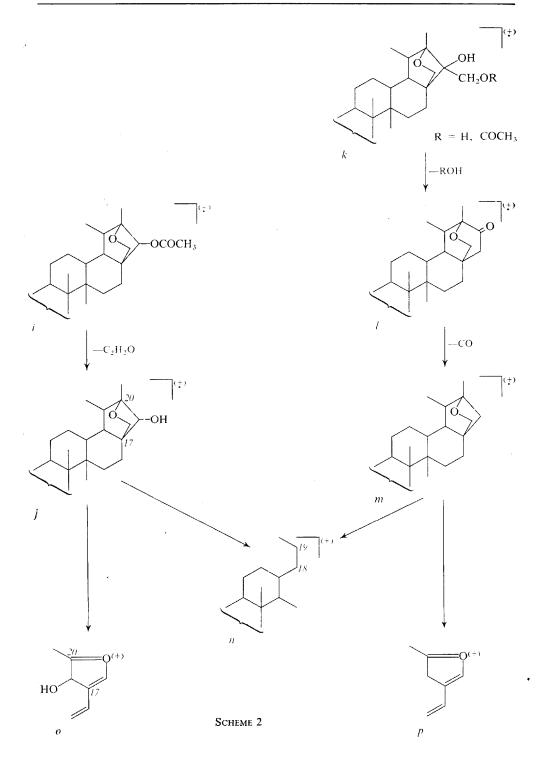
Fragmentation of Compounds of Type A(II - X)

The elimination of the group bridging the tetrahydropyran ring aims at the ion f which is fragmented both simply to the ion $h (m/e \ 43)$ and to the ion g by the elimination of C₄H₈O through McLafferty's rearrangement. The substituent in the

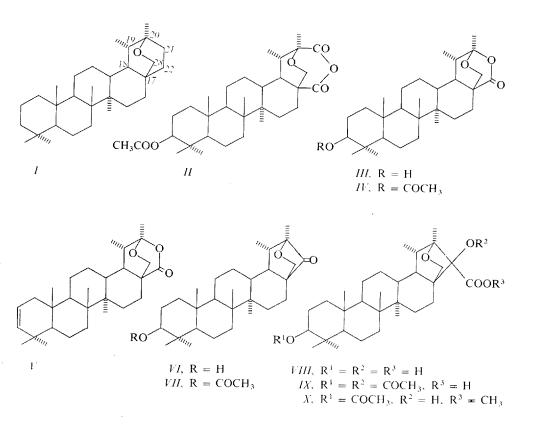
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position 3 β , together with the hydrogen from the neighbouring position, is eliminated as water or acetic acid at the level of fragments b, f and g. The elimination of water or acetic acid from the molecular ion was observed only with substances III, IV and X.

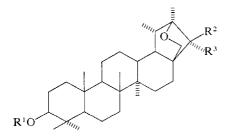


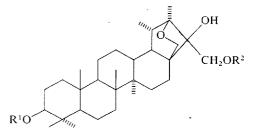
Fragmentation of Compounds of Type B(XI - XVIII)

The ion j or m is common to all substances of this type. It may represent the molecular ion or a fragment. It is cleaved at the tetrahydropyran ring and ring D under two hydrogen atoms transfer, giving rise to ions n and o or p. During this the charge on the oxygen heteroatom of ring E prevails and the ions o and p are by one order of magnitude more intensive than the ion n. The ion j as a fragment is formed from the ion i by elimination of CH_2CO . The ion m is formed from the ion k by elimination of $H_2O + CO$ or $CH_3COOH + CO$, from ion l by elimination of CO. The elimination of water or acetic acid from ring A is mostly restricted to the molecular ion. Compounds XIII and XV are exceptions in which elimination of acetic acid

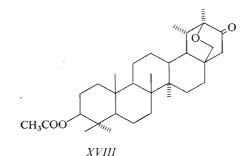
Collection Czechoslov, Chem, Commun. [Vol. 41] [1976]

takes place also from the fragment j, and also compound XVIII is an exception in which acetic acid is eliminated also from fragments m and n. The pair of the C₍₂₂₎ epimeric hydroxy derivatives XII and XIV and their acetyl derivatives XIII and XV have the same m/e values and approximately equal relative intensities of the corresponding ions in low-resolution records. Supposing that the ion l, formed from the ion k, possesses the same structure as the molecular ion of compound XVIII, the decomposition of the ion c may be visualized in an analogous manner. However, the ion d was not found in the spectra and the corresponding substance could not be prepared.





X1. $R^1 = R^2 = H$. $R^3 = OH$ X11. $R^1 = COCH_3$, $R^2 = H$. $R^3 = OH$ X111. $R^1 = COCH_3$, $R^2 = H$. $R^3 = OCOCH_3$ X1V. $R^1 = COCH_3$, $R^2 = OH$, $R^3 = H$ XV. $R^1 = COCH_3$, $R^2 = OCOCH_3$, R = H XVI, $\mathbf{R}^1 = \mathbf{R}^2 = \mathbf{H}$ XVII, $\mathbf{R}^1 = \mathbf{R}^2 = \text{COCH}_3$



The measurement of the metastable ions by the DADI (direct analysis of daughter ions) technique confirmed the $a \rightarrow b$ transition for substance II (type A), the sequence $b \rightarrow f \rightarrow g$ for substance V (type A), and the decomposition $l \rightarrow m \rightarrow n$ for compound XVIII (type B).

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2. Budzikiewicz H., Djerassi C., Williams D. H.: Structure Elucidation of Natural Products by Mass Spectrometry, Vol. 2, p. 137. Holden-Day, San Francisco, London, Amsterdam 1964.

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